5825 measured reflections

 $R_{\rm int} = 0.025$ 

3187 independent reflections

2422 reflections with  $I > 2\sigma(I)$ 

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# 4-(2-Furyl)-1-(4-methoxybenzyl)-1Hpyrazolo[3,4-d]pyrimidine

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Key indicators: single-crystal X-ray study; T = 112 K; mean  $\sigma$ (C–C) = 0.002 Å; R factor = 0.046; wR factor = 0.135; data-to-parameter ratio = 15.2.

The title compound, C<sub>17</sub>H<sub>14</sub>N<sub>4</sub>O<sub>2</sub>, is a close analog of potent antimycobacterial purines. Molecules are linked by C-H···N hydrogen bonds, forming infinite chains along [010]. The secondary structure is further stabilized by weak intermolecular contacts of types  $C-H \cdots N$  and  $C-H \cdots O$ .

#### **Related literature**

Most bond lengths and angles are in good agreement with those found for 6-furyl- and 6-thienyl-9-benzylpurines (Braendvang & Gundersen, 2007a; Mazumdar et al., 2001), except that the furyl ring in the title compound is rotated by approximately 180°. This orientation of the furyl ring is stabilized by  $C-H \cdots O$  hydrogen bonding.

For related literature, see: Bakkestuen et al. (2000, 2005); Brændvang & Gundersen (2005, 2007b); Gundersen et al. (2002).



#### **Experimental**

Crystal data

 $C_{17}H_{14}N_4O_2$  $M_r = 306.32$ Triclinic,  $P\overline{1}$ a = 8.2163 (15) Åb = 8.4765 (15) Å c = 12.437 (2) Å  $\alpha = 72.485(3)^{\circ}$  $\beta = 74.209 \ (3)^{\circ}$ 

 $\gamma = 61.065 \ (3)^{\circ}$ V = 714.6 (2) Å<sup>3</sup> Z = 2Mo  $K\alpha$  radiation  $\mu = 0.10 \text{ mm}^-$ T = 112 (2) K  $0.20 \times 0.20 \times 0.15~\text{mm}$ 

#### Data collection

Siemens SMART CCD diffractometer Absorption correction: multi-scan

(SADABS; Sheldrick, 1996)  $T_{\min} = 0.661, \ T_{\max} = 0.986$ 

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.046$	210 parameters
$wR(F^2) = 0.135$	H-atom parameters constrained
S = 1.03	$\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$
3187 reflections	$\Delta \rho_{\rm min} = -0.30 \text{ e } \text{\AA}^{-3}$

#### Table 1

Hydrogen-bond	geometry	(Å, °	)
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$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C21 - H21 \cdots N2^{i}$	0.95	2.41	3.338 (2)	164
$C16-H16A\cdots N7^{ii}$	0.98	2.60	3.572 (2)	174
C3−H3···O18	0.95	2.60	3.029 (2)	108
$C6 - H6 \cdots O15^{iii}$	0.95	2.51	3.344 (2)	147

Symmetry codes: (i) x, y - 1, z; (ii) x - 1, y + 1, z; (iii) -x + 1, -y + 1, -z + 2.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and POV-RAY for Windows (Persistence of Vision Pty. Ltd, 2004); software used to prepare material for publication: SHELXL97 and WinGX (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2248).

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supplementary materials

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## 4-(2-Furyl)-1-(4-methoxybenzyl)-1*H*-pyrazolo[3,4-*d*]pyrimidine

## M. Brændvang and L.-L. Gundersen

#### Comment

We have discovered that certain 6-aryl-9-benzylpurines are potent antimycobacterials *in vitro* and may have a potential as antitubercular drugs (Bakkestuen *et al.*, 2000; Gundersen *et al.*, 2002; Bakkestuen *et al.*, 2005; Braendvang & Gundersen, 2005). Compound I was synthesised as a non-purine analog of the previously reported antimycobacterial purines (Braendvang & Gundersen, 2007 b).

The molecular geometries are illustrated in Fig. 1 and the bond lengths and angles are listed in Table 1. The angle between the mean plane of the pyrazolo-pyrimidine ring system and the furyl ring are 0.66 (8)°. This coplanarity has also been reported for 6-arylpurines, except that the furyl ring in compound **I** is rotated approximately 180° compared to previously reported purine structure (Braendvang & Gundersen, 2007a). The orientation of the furyl group is stabilized by intramolecular C3–H3···O18 hydrogen bonding (Table 2). The benzene ring plane is inclined at an angle of 72.83 (7)° to the pyrazolo-pyrimidine ring system, and is very similar as in case of purine ring system previously reported (Braendvang & Gundersen, 2007a).

Fig. 2. shows the molecules in the unit cell. The molecular packing is similar to previous reported purine analog (Braend-vang & Gundersen, 2007a), except that the infinite chain along [010] is linked together by hydrogen bonding C21–H21···N2<sup>i</sup> [Table 2 and Fig. 2; symmetry code: (i) x, y-1, z] while in the previous reported structure the hydrogen bonding is between the oxygen in the furyl ring and the H8 in the purine ring.

### Experimental

The title compound was synthesized as described by Brændvang & Gundersen (2007 b). Crystals suitable for X-ray diffraction studies were obtained by slow evaporation at room temperature from a mixture of acetone and hexane (1:2).

#### Refinement

H atoms were positioned geometrically and allowed to ride and rotate (for the CH<sub>3</sub> group) on their carrier atoms, with C—H bond lengths of 0.95 (aromatic C—H), 0.99 (CH<sub>2</sub>) or 0.98 Å (CH<sub>3</sub>) and with  $U_{iso}(H) = 1.2U_{eq}(C)$  for methylene and aromatic C—H or  $1.5U_{eq}(C)$  for CH<sub>3</sub>.

Figures



Fig. 1. The molecular structure of (I), with atom labels and 50% probability displacement ellipsoids for non-H atoms.



Fig. 2. The packing of (I), viewed down the b axis, showing the infinite chains of (I) along baxis. H atoms have been omitted for clarity.

## 4-(2-Furyl)-1-(4-methoxybenzyl)-1H-pyrazolo[3,4,d]pyrimidine

Crystal data	
$C_{17}H_{14}N_4O_2$	<i>Z</i> = 2
$M_r = 306.32$	$F_{000} = 320$
Triclinic, <i>P</i> T	$D_{\rm x} = 1.424 {\rm ~Mg~m^{-3}}$
Hall symbol: -P 1	Mo <i>K</i> $\alpha$ radiation $\lambda = 0.71073$ Å
<i>a</i> = 8.2163 (15) Å	Cell parameters from 2969 reflections
b = 8.4765 (15)  Å	$\theta = 2.8 - 28.3^{\circ}$
c = 12.437 (2)  Å	$\mu = 0.10 \text{ mm}^{-1}$
$\alpha = 72.485 \ (3)^{\circ}$	T = 112 (2)  K
$\beta = 74.209 \ (3)^{\circ}$	Block, yellow
$\gamma = 61.065 \ (3)^{\circ}$	$0.20\times0.20\times0.15~mm$
V = 714.6 (2) Å <sup>3</sup>	

#### Data collection

Siemens SMART CCD diffractometer	$R_{\rm int} = 0.025$
ω scans	$\theta_{max} = 28.8^{\circ}$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$\theta_{\min} = 1.7^{\circ}$
$T_{\min} = 0.661, \ T_{\max} = 0.986$	$h = -10 \rightarrow 9$
5825 measured reflections	$k = -10 \rightarrow 11$
3187 independent reflections	$l = -15 \rightarrow 16$
2422 reflections with $I > 2\sigma(I)$	

Refinement

Refinement on  $F^2$ Least-squares matrix: full  $R[F^2 > 2\sigma(F^2)] = 0.046$  $wR(F^2) = 0.135$ S = 1.033187 reflections 210 parameters H-atom parameters constrained  $w = 1/[\sigma^2(F_0^2) + (0.0927P)^2]$ where  $P = (F_0^2 + 2F_c^2)/3$  
$$\begin{split} &(\Delta/\sigma)_{max} < 0.001 \\ &\Delta\rho_{max} = 0.29 \text{ e } \text{ Å}^{-3} \\ &\Delta\rho_{min} = -0.30 \text{ e } \text{ Å}^{-3} \\ &\text{Extinction correction: SHELXL97,} \\ &\text{Fc}^* = \text{kFc}[1 + 0.001 \text{xFc}^2 \lambda^3 / \sin(2\theta)]^{-1/4} \\ &\text{Extinction coefficient: } 0.009 (5) \end{split}$$

### Special details

**Geometry**. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

	x	у	Ζ	$U_{\rm iso}*/U_{\rm eq}$
C16	0.0212 (2)	1.2299 (2)	1.00299 (13)	0.0310 (4)
H16B	-0.0885	1.2588	1.0631	0.046*
H16A	-0.0199	1.2561	0.93	0.046*
H16C	0.0847	1.3052	0.9974	0.046*
C12	0.3085 (2)	0.9750 (2)	0.95405 (12)	0.0219 (3)
C13	0.3569 (2)	1.0823 (2)	0.85326 (12)	0.0246 (3)
H13	0.2734	1.2083	0.8316	0.03*
C14	0.5284 (2)	1.0031 (2)	0.78483 (12)	0.0243 (3)
H14	0.5622	1.0772	0.7172	0.029*
С9	0.6511 (2)	0.8187 (2)	0.81324 (12)	0.0227 (3)
C10	0.5980 (2)	0.7109 (2)	0.91305 (12)	0.0249 (3)
H10	0.6788	0.5835	0.933	0.03*
C11	0.4287 (2)	0.7889 (2)	0.98257 (12)	0.0254 (3)
H11	0.3944	0.7149	1.0501	0.03*
C8	0.8340 (2)	0.7353 (2)	0.73554 (13)	0.0277 (4)
H8B	0.8838	0.8279	0.7005	0.033*
H8A	0.9272	0.6296	0.7814	0.033*
C3	0.7299 (2)	0.7047 (2)	0.48624 (12)	0.0238 (3)
H3	0.6905	0.7563	0.4133	0.029*
C3A	0.7731 (2)	0.5197 (2)	0.54462 (11)	0.0199 (3)
C4	0.7772 (2)	0.3590 (2)	0.53073 (11)	0.0201 (3)
C6	0.8695 (2)	0.2165 (2)	0.70764 (12)	0.0277 (4)
H6	0.9036	0.1068	0.7644	0.033*

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(A^2)$ 

# supplementary materials

C7A	0.8228 (2)	0.5069 (2)	0.64762 (12)	0.0212 (3)
C17	0.7315 (2)	0.3450 (2)	0.43035 (11)	0.0209 (3)
C19	0.6430 (2)	0.4520 (2)	0.26199 (13)	0.0283 (4)
H19	0.6031	0.5326	0.1925	0.034*
C20	0.6719 (2)	0.2745 (2)	0.29123 (12)	0.0279 (4)
H20	0.6569	0.2094	0.2473	0.033*
C21	0.7291 (2)	0.2044 (2)	0.40082 (12)	0.0249 (3)
H21	0.7595	0.0833	0.4447	0.03*
N1	0.81128 (18)	0.67308 (17)	0.64542 (10)	0.0239 (3)
N2	0.75177 (19)	0.79507 (17)	0.54690 (10)	0.0260 (3)
N5	0.82546 (19)	0.20712 (17)	0.61400 (10)	0.0251 (3)
N7	0.87220 (19)	0.35803 (18)	0.73238 (10)	0.0256 (3)
015	0.14786 (15)	1.04026 (15)	1.02991 (8)	0.0279 (3)
O18	0.67872 (15)	0.50055 (14)	0.34530 (8)	0.0254 (3)

Atomic displacement parameters  $(\text{\AA}^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C16	0.0294 (9)	0.0345 (9)	0.0258 (8)	-0.0120 (8)	-0.0011 (7)	-0.0081 (7)
C12	0.0244 (8)	0.0289 (8)	0.0170 (6)	-0.0158 (7)	-0.0021 (6)	-0.0044 (6)
C13	0.0308 (9)	0.0220 (8)	0.0221 (7)	-0.0144 (7)	-0.0035 (6)	-0.0012 (6)
C14	0.0331 (9)	0.0273 (8)	0.0190 (7)	-0.0208 (7)	-0.0018 (6)	-0.0018 (6)
C9	0.0267 (8)	0.0288 (8)	0.0210 (7)	-0.0172 (7)	-0.0035 (6)	-0.0075 (6)
C10	0.0295 (8)	0.0226 (8)	0.0240 (7)	-0.0118 (7)	-0.0087 (6)	-0.0015 (6)
C11	0.0316 (9)	0.0293 (8)	0.0175 (6)	-0.0183 (7)	-0.0061 (6)	0.0030 (6)
C8	0.0272 (8)	0.0357 (9)	0.0288 (8)	-0.0184 (7)	-0.0011 (6)	-0.0128 (7)
C3	0.0287 (8)	0.0240 (8)	0.0193 (7)	-0.0152 (7)	-0.0024 (6)	0.0000 (6)
C3A	0.0218 (7)	0.0218 (7)	0.0161 (6)	-0.0121 (6)	-0.0004 (5)	-0.0017 (5)
C4	0.0224 (7)	0.0216 (7)	0.0168 (6)	-0.0130 (6)	0.0002 (5)	-0.0012 (5)
C6	0.0370 (9)	0.0261 (8)	0.0181 (7)	-0.0160 (7)	-0.0045 (6)	0.0023 (6)
C7A	0.0225 (8)	0.0243 (8)	0.0183 (7)	-0.0133 (6)	0.0010 (6)	-0.0044 (6)
C17	0.0225 (8)	0.0230 (8)	0.0172 (6)	-0.0135 (6)	-0.0009 (6)	0.0000 (6)
C19	0.0331 (9)	0.0377 (9)	0.0190 (7)	-0.0202 (8)	-0.0055 (6)	-0.0028 (6)
C20	0.0329 (9)	0.0377 (9)	0.0223 (7)	-0.0229 (8)	-0.0005 (6)	-0.0088 (7)
C21	0.0300 (8)	0.0260 (8)	0.0220 (7)	-0.0169 (7)	-0.0008 (6)	-0.0037 (6)
N1	0.0294 (7)	0.0250 (7)	0.0210 (6)	-0.0157 (6)	-0.0024 (5)	-0.0044 (5)
N2	0.0321 (7)	0.0233 (7)	0.0239 (6)	-0.0159 (6)	-0.0029 (5)	-0.0012 (5)
N5	0.0339 (7)	0.0239 (7)	0.0184 (6)	-0.0166 (6)	-0.0032 (5)	0.0009 (5)
N7	0.0327 (7)	0.0269 (7)	0.0176 (6)	-0.0151 (6)	-0.0046 (5)	-0.0007 (5)
015	0.0262 (6)	0.0327 (6)	0.0193 (5)	-0.0124 (5)	-0.0004 (4)	-0.0016 (4)
O18	0.0357 (6)	0.0242 (6)	0.0185 (5)	-0.0168 (5)	-0.0071 (4)	0.0019 (4)

Geometric parameters	(Å,	°)
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C16—O15	1.423 (2)	C3—N2	1.3199 (19)
C16—H16B	0.98	С3—С3А	1.425 (2)
C16—H16A	0.98	С3—Н3	0.95
C16—H16C	0.98	C3A—C7A	1.4063 (19)
C12—O15	1.3669 (18)	C3A—C4	1.407 (2)

C12—C11	1.392 (2)	C4—N5	1.3470 (18)
C12—C13	1.396 (2)	C6—N7	1.336 (2)
C13—C14	1.392 (2)	C6—N5	1.344 (2)
С13—Н13	0.95	С6—Н6	0.95
C14—C9	1.386 (2)	C7A—N7	1.3437 (18)
C14—H14	0.95	C7A—N1	1.3574 (18)
C9—C10	1.405 (2)	C17—C21	1.358 (2)
N1—C8	1.4610 (18)	C17—O18	1.3793 (16)
C4—C17	1.4494 (19)	C19—C20	1.351 (2)
C8—C9	1.512 (2)	C19—O18	1.3667 (18)
C10-C11	1.385 (2)	С19—Н19	0.95
C10—H10	0.95	C20—C21	1.422 (2)
C11—H11	0.95	C20—H20	0.95
C8—H8B	0.99	C21—H21	0.95
C8—H8A	0.99	N1—N2	1.3726 (17)
O15-C16-H16B	109.5	C7A—C3A—C4	115.70 (13)
O15—C16—H16A	109.5	C7A—C3A—C3	104.23 (12)
H16B—C16—H16A	109.5	C4—C3A—C3	140.06 (13)
O15—C16—H16C	109.5	N5-C4-C3A	119.68 (13)
H16B—C16—H16C	109.5	N5-C4-C17	116.53 (13)
H16A—C16—H16C	109.5	C3A—C4—C17	123.80 (13)
O15—C12—C11	115.94 (13)	N7—C6—N5	129.32 (14)
O15-C12-C13	124.37 (14)	N7—C6—H6	115.3
C11—C12—C13	119.68 (14)	N5—C6—H6	115.3
C14—C13—C12	119.47 (14)	N7—C7A—N1	126.44 (13)
C14—C13—H13	120.3	N7—C7A—C3A	126.35 (13)
С12—С13—Н13	120.3	N1—C7A—C3A	107.22 (13)
C9—C14—C13	121.45 (14)	C21—C17—O18	110.44 (12)
C9—C14—H14	119.3	C21—C17—C4	132.49 (14)
C13—C14—H14	119.3	O18—C17—C4	117.07 (12)
C14—C9—C10	118.51 (14)	C20-C19-O18	111.19 (13)
C14—C9—C8	120.25 (14)	С20—С19—Н19	124.4
C10—C9—C8	121.19 (14)	O18—C19—H19	124.4
C11—C10—C9	120.49 (15)	C19—C20—C21	106.41 (13)
C11-C10-H10	119.8	С19—С20—Н20	126.8
С9—С10—Н10	119.8	C21—C20—H20	126.8
C10-C11-C12	120.35 (14)	C17—C21—C20	106.38 (13)
C10-C11-H11	119.8	C17—C21—H21	126.8
C12—C11—H11	119.8	C20-C21-H21	126.8
N1—C8—C9	112.29 (12)	C7A—N1—N2	110.83 (12)
N1—C8—H8B	109.1	C7A—N1—C8	128.64 (13)
С9—С8—Н8В	109.1	N2—N1—C8	120.17 (12)
N1—C8—H8A	109.1	C3—N2—N1	106.67 (12)
С9—С8—Н8А	109.1	C6—N5—C4	117.58 (13)
H8B—C8—H8A	107.9	C6—N7—C7A	111.38 (12)
N2—C3—C3A	111.02 (13)	C12—O15—C16	117.41 (12)
N2—C3—H3	124.5	C19—O18—C17	105.58 (11)
СЗА—СЗ—НЗ	124.5		

# supplementary materials

O15-C12-C13-C14	177.07 (13)	C3A—C4—C17—O18	-1.0 (2)
C11-C12-C13-C14	-2.6 (2)	O18—C19—C20—C21	0.29 (18)
C12—C13—C14—C9	1.4 (2)	O18—C17—C21—C20	0.04 (17)
C13-C14-C9-C10	0.6 (2)	C4—C17—C21—C20	179.89 (15)
C13—C14—C9—C8	178.30 (13)	C19—C20—C21—C17	-0.20 (17)
C14—C9—C10—C11	-1.5 (2)	N7—C7A—N1—N2	-178.42 (13)
C8—C9—C10—C11	-179.15 (13)	C3A—C7A—N1—N2	1.53 (16)
C9—C10—C11—C12	0.3 (2)	N7—C7A—N1—C8	-5.4 (2)
O15-C12-C11-C10	-177.94 (12)	C3A-C7A-N1-C8	174.58 (14)
C13-C12-C11-C10	1.7 (2)	C9—C8—N1—C7A	-88.40 (19)
C14—C9—C8—N1	-85.88 (17)	C9—C8—N1—N2	84.09 (17)
C10-C9-C8-N1	91.74 (17)	C3A—C3—N2—N1	0.61 (17)
N2—C3—C3A—C7A	0.28 (17)	C7A—N1—N2—C3	-1.34 (16)
N2-C3-C3A-C4	179.00 (18)	C8—N1—N2—C3	-175.07 (13)
C7A—C3A—C4—N5	0.4 (2)	N7—C6—N5—C4	0.4 (3)
C3—C3A—C4—N5	-178.22 (17)	C3A-C4-N5-C6	-0.5 (2)
C7A—C3A—C4—C17	-179.69 (13)	C17—C4—N5—C6	179.62 (13)
C3—C3A—C4—C17	1.7 (3)	N5—C6—N7—C7A	-0.2 (2)
C4—C3A—C7A—N7	-0.2 (2)	N1—C7A—N7—C6	-179.97 (14)
C3—C3A—C7A—N7	178.87 (14)	C3A—C7A—N7—C6	0.1 (2)
C4—C3A—C7A—N1	179.84 (12)	C11-C12-O15-C16	-179.58 (13)
C3—C3A—C7A—N1	-1.08 (16)	C13—C12—O15—C16	0.8 (2)
N5-C4-C17-C21	-0.9 (2)	C20-C19-O18-C17	-0.27 (17)
C3A—C4—C17—C21	179.19 (15)	C21-C17-O18-C19	0.13 (16)
N5-C4-C17-O18	178.94 (12)	C4-C17-O18-C19	-179.74 (13)

## Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A
C21— $H21$ ···N2 <sup>i</sup>	0.95	2.41	3.338 (2)	164
C16—H16A···N7 <sup>ii</sup>	0.98	2.60	3.572 (2)	174
С3—Н3…О18	0.95	2.60	3.029 (2)	108
C6—H6···O15 <sup>iii</sup>	0.95	2.51	3.344 (2)	147

Symmetry codes: (i) *x*, *y*–1, *z*; (ii) *x*–1, *y*+1, *z*; (iii) –*x*+1, –*y*+1, –*z*+2.





